LAMS--1099

"NCLASSITIE OFFICIAL USE ONLY

Series--A

For Reference

C.3

NOT TO BE TAKEN FROM THIS ROOM

UNCLASSIFIED

LOS ALAMOS SCIENTIFIC LABORATORY

OF THE

UNIVERSITY OF CALIFORNIA

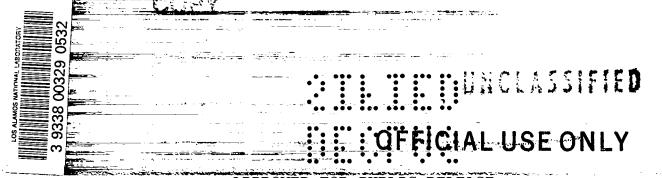
CONTRACT W-7405-ENG. 36 WITH

U. S. ATOMIC ENERGY COMMISSION

NO LO DE COMECTION ACON CONTRACTOR ACON COMECTION A

UNLLASSITIED

CIC-14 REPORT COLLECTION







PUBLICLY RELEASABLE

Per MONU . F35-16 Date: 4-25-95 By Kolar . CIC-14 Date: 8-12-96

LAMS- 1099

Series A

April 6, 1950

This document contains 8 pages.

PRELIMINARY REPORT ON HUMAN EXCRETION OF TRITIUM

Work Done By:

Ernest C. Anderson

Ernest Pinson

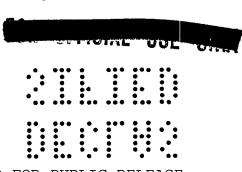
Written By:

Classification changed to UNCLASSIFIED Commission. Ernest C. Anderson by authority of the U. S. Atom. 1/3 4) Classification changed to UNCLASSIFIED By REPORT LIBRARY m. allew 1-14-5-7

UNCLASSIFIED

HEALTH AND BIOLOGY





APPROVED FOR PUBLIC RELEASE

LAMS - 1099

HEALTH AND BIOLOGY

	COPY NO.
Los Alamos	1-20
STANDARD DISTRIBUTION	
Argonne National Laboratory	21-28
Armed Forces Special Weapons Project	29
Atomic Energy Commission, Washington	30-31
Battelle Memorial Institute	32
Brookhaven National Laboratory	33-36
Bureau of Medicine and Surgery	37
Carbide and Carbon Chemicals Corporation (K-25 Plant)	38-41
Carbide and Carbon Chemicals Corporation (Y-12 Plant)	42-45
Chicago Operations Office	46
Columbia University (Dunning)	47
Columbia University (Failla)	48
General Electric Company, Richland	49-54
Idaho Operations Office	55-56
Iowa State College	57
Knolls Atomic Power Laboratory	58-59
Mallinckrodt Chemical Works	60
Massachusetts Institute of Technology (Kaufmann)	61
Mound Laboratory	62-64
National Advisory Committee for Aeronautics	65
National Bureau of Standards	66
Naval Radiological Defense Laboratory	67
NEPA Project	68
New Brunswick Laboratory	69
New York Operations Office	70-72
North American Aviation, Inc.	73
Oak Ridge National Laboratory	74-85
Patent Branch, Washington	86
Public Health Service	87
Sandia Laboratory	88
Sylvania Electric Products, Inc.	89
Technical Information Branch, ORE	90-104
UCLA Medical Research Laboratory (Warren)	105
University of California Radiation Laboratory	106 -110
University of Chicago Toxicity Laboratory	111
University of Rochester	112-113
University of Washington	114
Western Reserve University (Friedell	115-118
Westinghouse Electric Corporation	119

PRELIMINARY REPORT ON HUMAN EXCRETION OF TRITEUM

Ernest C. Anderson and Ernest Pinson

An investigation of the human body water turn-over and of the distribution and exchange equilibria of body water has been begun at this laboratory using tritium tracer. In this paper are reported some preliminary data which have been obtained on the rate of turn-over of body water.

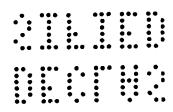
Six subjects received a few millicuries of tritlum by inhalation of isotopically labeled hydrogen gas. The concentration of H3 in the urine of these individuals has been followed for a period of some 15 days. Analyses were performed by allowing vapor from the urine to pass over hot Zn (zinc dust suspended on glass wool) in an evacuated system. The Ho produced by the reaction - H₂O + Zn - H₂ + ZnO - was passed through a dry ice-cooled trap to remove any unreacted water and other chemical species of low volatility. The non-condensable gases were transferred to an ion chamber by means of a Toeppler pump. Urine samples of 0.1 cc. volume were reduced giving a pressure of 412 mm. Hg in a 250 cc. chamber. The pressure was increased to 970 mm. with carbon dioxide and the ion current measured with a vibrating-reed electrometer. Relative values only are given in this report. Ion currents of the order of 10 to 100 times background were obtained from the samples. Only negligible amounts of material were found condensed in the dry-ice trap and quantitative measurements with pure water indicate that the reduction procedure is stoichiometric to better than 95%.

The rate of excretion of the tritium was found to be constant for a given subject but to vary considerably among individuals. Data on five individuals



arbitrarily normalized to coincide at zero time are plotted in Figure 1, showing a range in biological half-life from about 9 days to nearly 13 days. These values are to be compared with the value calculated for the Chalk River "Standard Man" of 13.5 days, assuming the tritium to be confined to the body water.

Figure 2 shows the results of a more detailed experiment on a sixth subject. The daily total volume of urine excreted by this individual was measured for a period of 6 days on a "normal" but uncontrolled water intake. From the sixth to the tenth day inclusive, the subject consumed as large a quantity of water as was reasonable during the course of his normal activities. Urine volume was measured daily during this period also. On the eleventh day, the subject returned to his "normal" water intake. During the first period, an average urine output of 1.1 liters per day was observed. This value rose to 5.0 liters per day during the middle period. Half-times of 12.5 days, 4.8 days and roughly 14.3 days were observed during these periods. The abrupt increases in excretion rate observed twice during the third period make it difficult to interpret this portion of the data. It is believed that these periods of increased elimination can be correlated with periods of unusual activity and increased water consumption. Assuming that the total body water and the total water elimination in forms other than the urine remained constant, it is possible to estimate the effective size of the reservoir from which H3 is being eliminated. Using only the first and second periods and letting V1 = volume of water eliminated in the urine per day; V2 = volume of water eliminated in other forms per day during the



first period, the primed symbols being the corresponding values for the second period; and V_3 = total volume of the H^3 reservoir expressed as water equivalents, we have:

For the first period:
$$\frac{V_1 + V_2}{V_3} = .055 \text{ day}^{-1}$$
and for the second period:
$$\frac{V_1^* + V_2^*}{3} = 0.144 \text{ day}^{-1}$$

$$V_1^* = 1.08 \text{ liters per day}$$

$$V_1^* = 5.0 \text{ liters per day}$$

$$V_2 = V_2^* \text{ and: } V_3 = V_3^*$$
we find that:
$$V_2 = 1.38 \text{ liters per day}$$

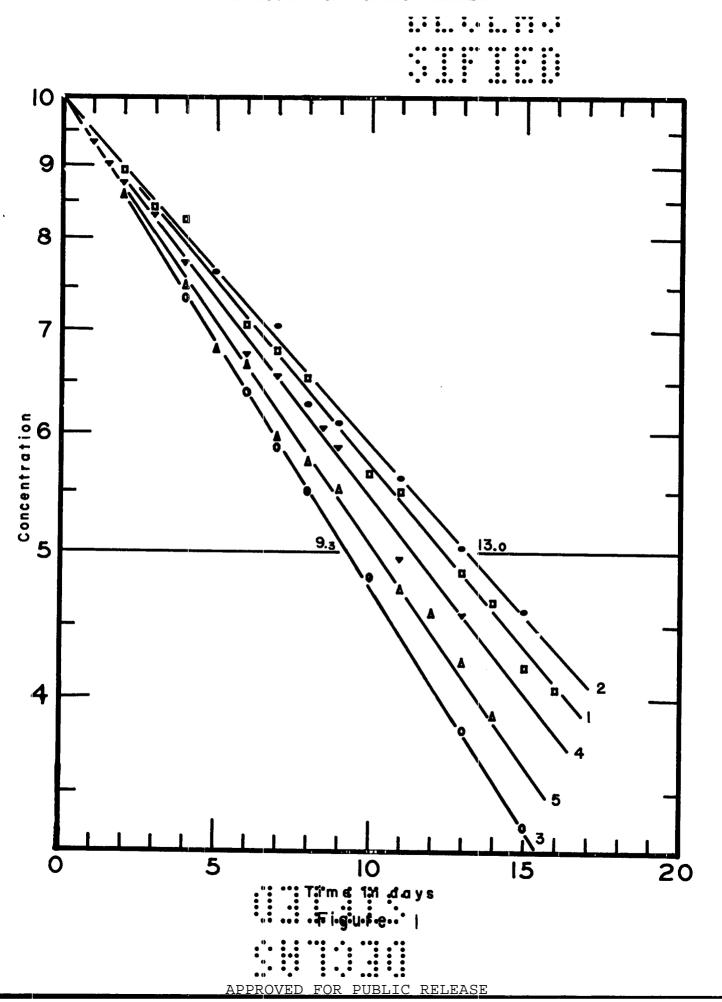
$$V_3 = 44 \text{ kg.}$$

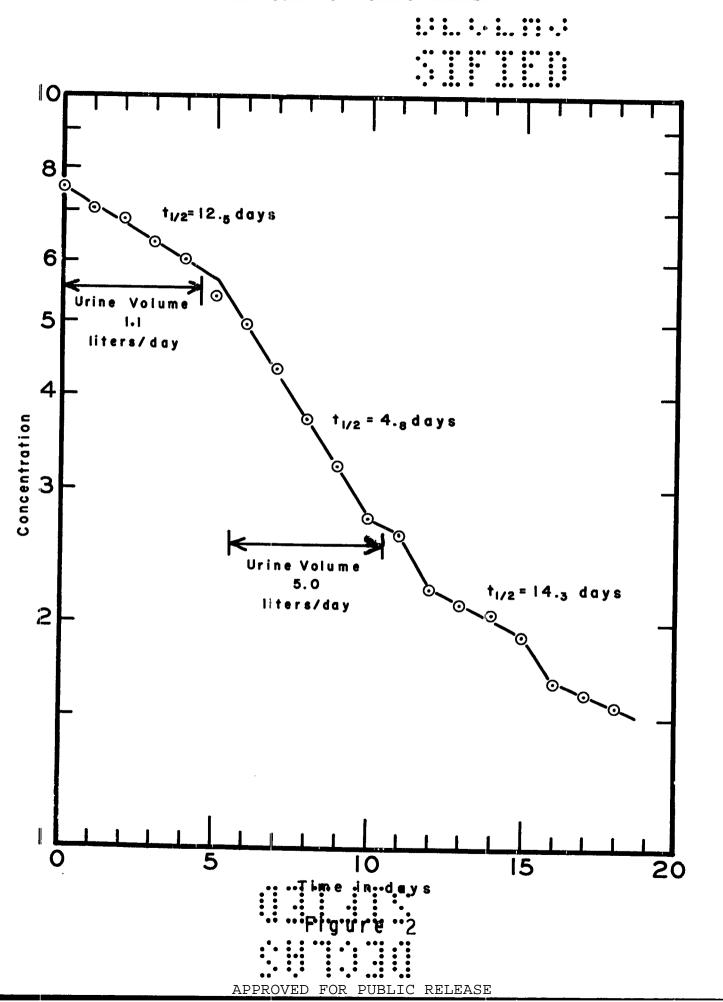
Since Subject #5 has a body weight of 65 kg. the "reservoir" corresponds to 68 per cent of total body weight in agreement with the assumption that the reservoir is principally body water with no large degree of exchange with other hydrogen, although it must be admitted that this calculation is not a sensitive test of the assumption. It is hoped that further data on this point will be obtained in future experiments.

It is planned to measure the H³ content of blood plasma, exhaled water wapor and perspiration and to look for any gradual lengthening of the biological half-time with time after exposure which might indicate a slow, reversible exchange with other chemical forms of hydrogen in the body. Animal experiments permitting detailed examination of specific body organs and chemicals for tritium uptake are also planned.



exchanges only with the water of the body, an assumption on which it is hoped to obtain some relevant information in the course of these experiments, it is possible to calculate a value for the amount of tritium which can safely be tolerated in the body. Assuming irradiation of the body by tritium in body fluids to be equivalent to an external irradiation by penetrating X- or gamma rays and using the accepted tolerance of 0.3 rep/week for this type of irradiation, ten millicuries is found to be the maximum permissible body tolerance for tritium. This value differs from the one millicurie value recommended at Chalk River by the arbitrary additional safety factor of ten which the committee elected to introduce.





DOCUMENT ROOM

REC. FROM Lug-1

DATE 5-5-50

REC.____NO. REC.____